

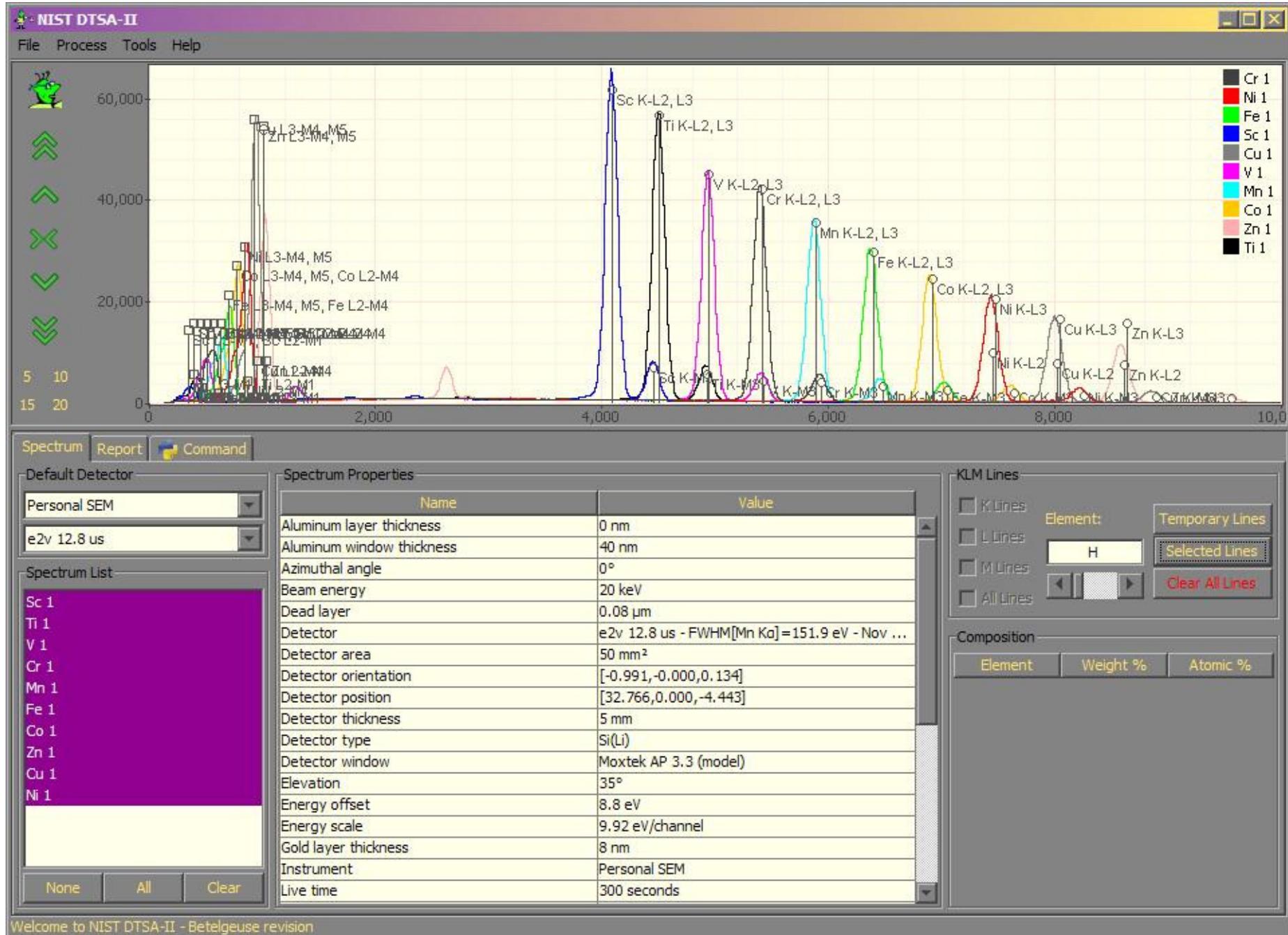
# BASIC SCRIPTING IN DTSA-II

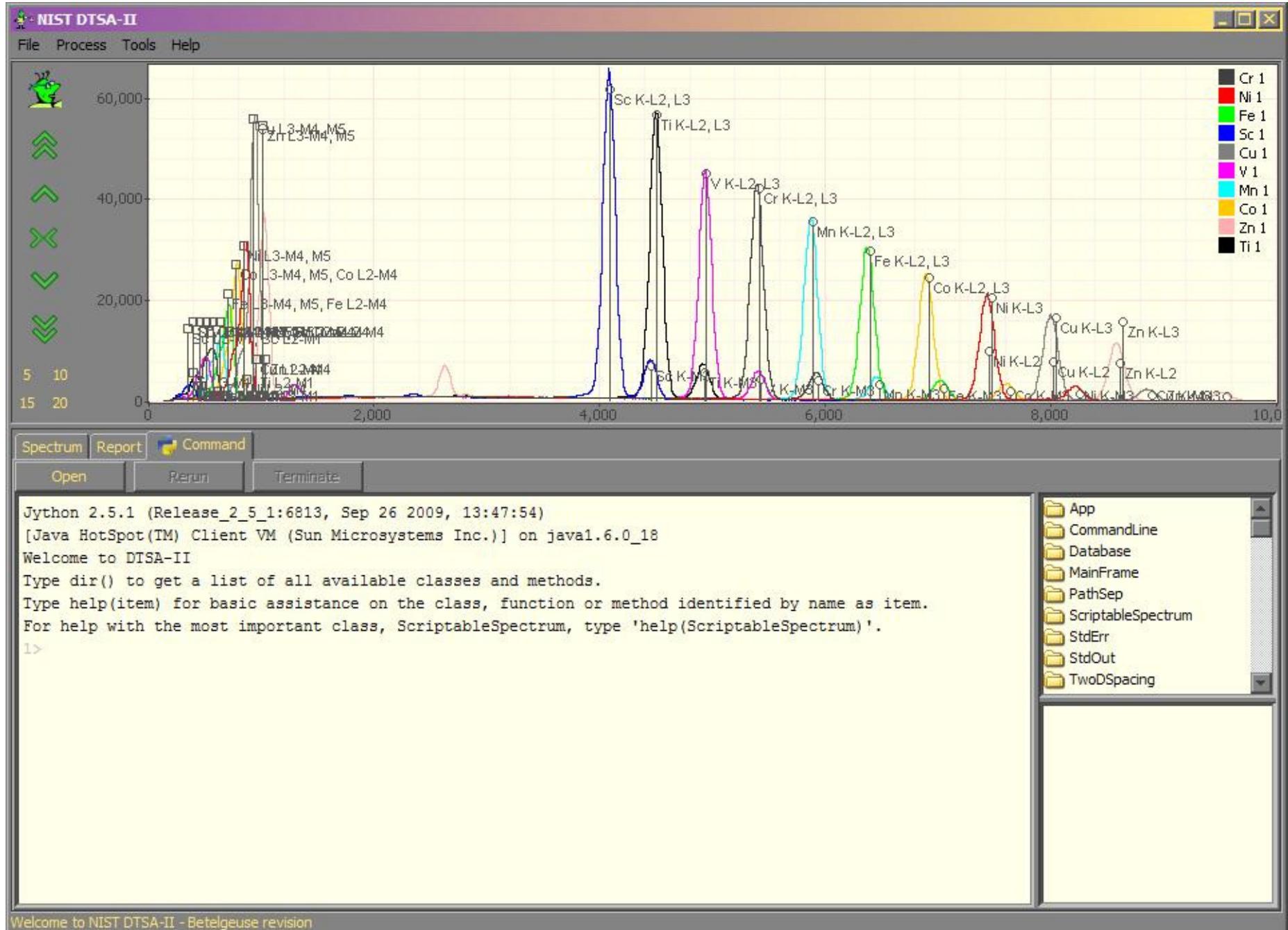
NICHOLAS W. M. RITCHIE

NATIONAL INSTITUTE OF STANDARDS AND TECHNOLOGY  
GAIITHERSBURG, MD 20899-8372



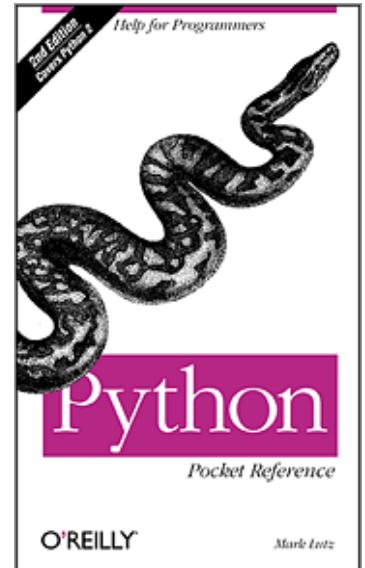
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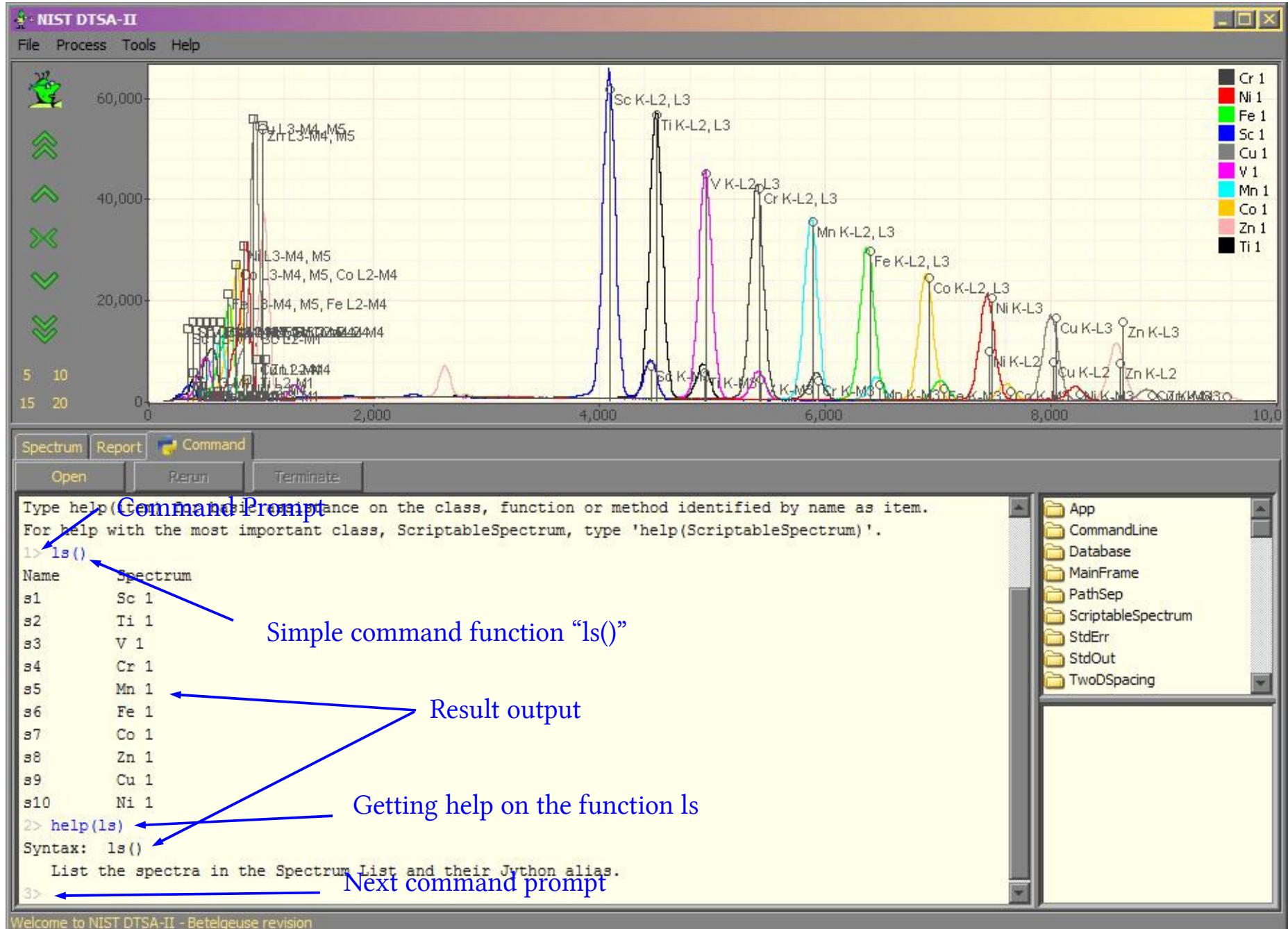


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# OVERVIEW



- Python 2.5 Syntax Scripting
  - General purpose, object oriented programming language
- Libraries
  - Access to all Java libraries
  - Access to all DTSA-II quantitative algorithms
  - Access to other algorithms not available through the DTSA-II GUI



# USEFUL UTILITY COMMANDS

## Generic Python Utility Methods

- dir() Displays a list of global functions, classes and variables
- dir(obj) Displays a list of methods and properties of the object *obj*
- help() Displays global help
- help(obj) Displays help on the object *obj*

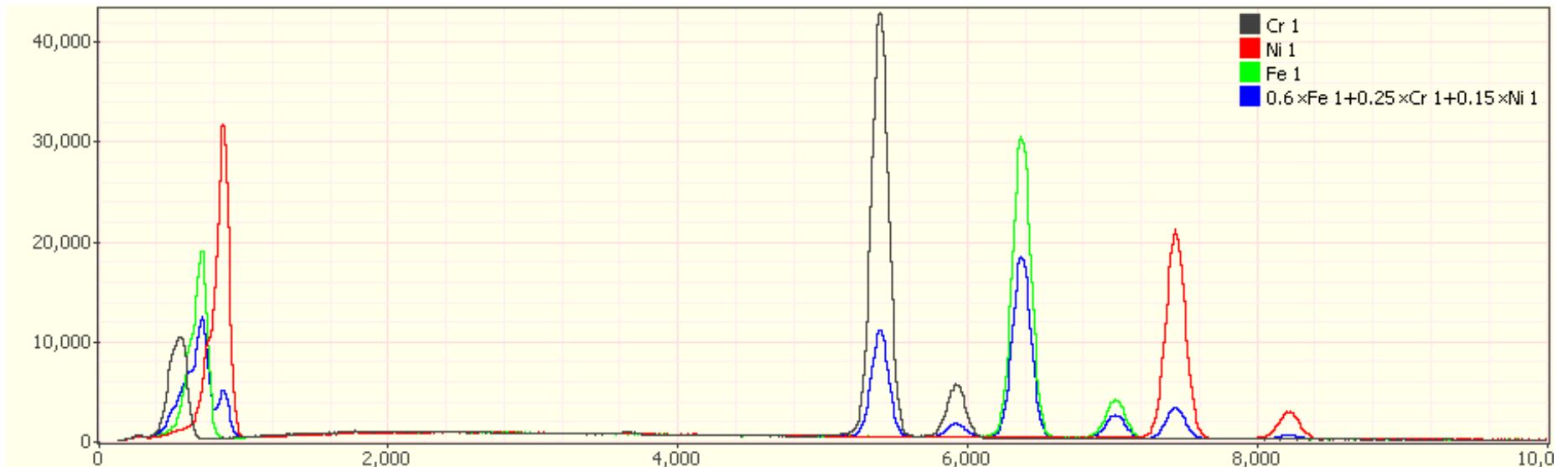
## DTSA-II Specific Utility Methods

- ls() Displays a list of all spectra in the Spectrum List with aliases ('s?' where ? is an integer)
- spectra() Returns a list object containing the spectra in the Spectrum List. spectra(True) returns only the selected spectra.
- listDetectors() Displays a list of all available detectors with aliases ('d?' where ? is an integer)
- getElement() GUI to select a single element
- getElements() GUI to select multiple elements
- createMaterial() GUI to select or create a new Composition or Material



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# SPECTRUM MATH



Assign the result  
to the variable "res"

Construct a spectrum that is  
 $0.6 \cdot \text{Fe 1} + 0.25 \cdot \text{Cr 1} + 0.15 \cdot \text{Ni 1}$

3> res=(0.6\*s6+0.25\*s4+0.15\*s10)  
4> res.display()

- You don't need to declare variables.
- Variable type is defined by what it contains.

Show the spectrum in the DTSA-II spectrum display

Operations: +, -, \*, abs()



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Command “dir(res)” returns a list of the names of the “methods” available for use with the contents of the variable “res”

3> res=(0.6\*s6+0.25\*s4+0.15\*s10)

4> res.display()

5> dir(res)

['NullSpectrum', '\_\_add\_\_', '\_\_class\_\_', '\_\_delattr\_\_', '\_\_dict\_\_', '\_\_doc\_\_', '\_\_eq\_\_', '\_\_ge\_\_', '\_\_getattribute\_\_', '\_\_getitem\_\_', '\_\_getslice\_\_', '\_\_gt\_\_', '\_\_hash\_\_', '\_\_initProxy\_\_', '\_\_init\_\_', '\_\_le\_\_', '\_\_len\_\_', '\_\_lt\_\_', '\_\_module\_\_', '\_\_mul\_\_', '\_\_ne\_\_', '\_\_new\_\_', '\_\_reduce\_\_', '\_\_reduce\_ex\_\_', '\_\_repr\_\_', '\_\_rmul\_\_', '\_\_setattr\_\_', '\_\_str\_\_', '\_\_sub\_\_', '\_\_supernames\_\_', '\_\_weakref\_\_', '\_getPyInstance', '\_getPySystemState', '\_setPyInstanceState', '\_setPySystemState', 'abs', 'applyLLT', 'autoOffset', 'backgroundCorrect', 'beamEnergy', 'channel', 'channelCount', 'channelWidth', 'class', 'classDictInit', 'clone', 'compareTo', 'composition', 'display', 'duaneHunt', 'energies', 'energy', 'equals', 'finalize', 'firstNonZeroChannel', 'fromXML', 'getChannelCount', 'getChannelWidth', 'getClass', 'getCounts', 'getFWHMAtMnKa', 'getProperties', 'getWrapped', 'getZeroOffset', 'hashCode', 'initializeSpectrumIndex', 'lastNonZeroChannel', 'liveTime', 'maxChannel', 'notify', 'notifyAll', 'offset', 'partition', 'peakIntegral', 'positiveDefinite', 'probeCurrent', 'properties', 'property', 'remap', 'rename', 'save', 'scale', 'setAsStandard', 'setEnergyScale', 'setFWHMAtMnKa', 'setLiveTime', 'setProbeCurrent', 'smooth', 'subSample', 'super\_\_toString', 'super\_\_toXML', 'takeOffAngle', 'toDouble', 'toString', 'toXML', 'totalCounts', 'wait', 'wrapped', 'zeroOffset']

6> res.getChannelWidth()

9.924458080007994

7> help(res.smooth)

Syntax: s.smooth()

Performs a fifth-order Savitzky-Golay filter on the spectrum s.

Getting help on a method

Calling convention is:

var.method()

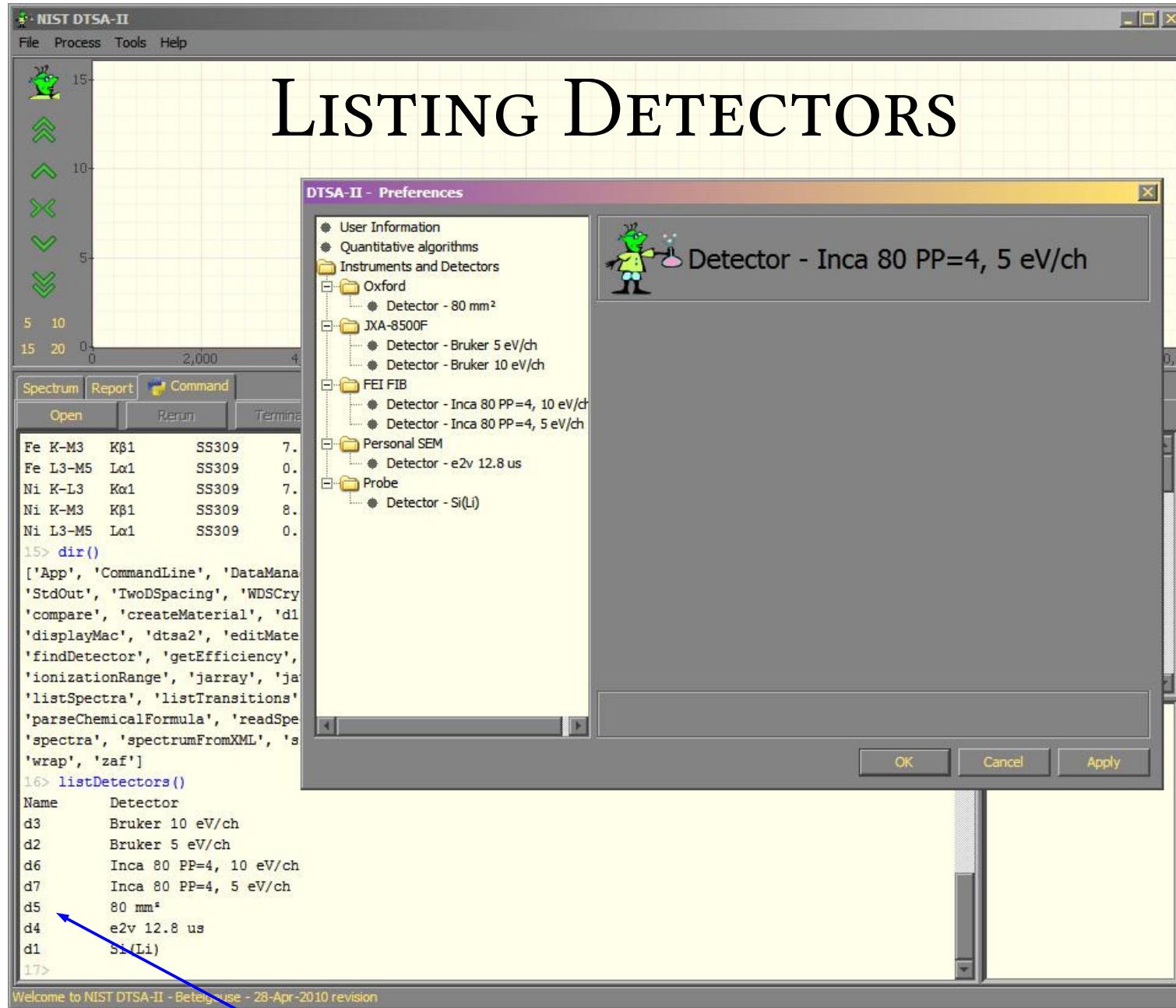
Get the string representation of “res”  
and call it “oldName”

12> oldName=str(res)  
13> oldName  
'Sum[0.6\xd7Fe 1+0.25\xd7Cr 1+0.15\xd7Ni 1]'  
14> res.rename("Bogus")  
15> res.rename(oldName)

Give the spectrum the name “Bogus”

Set it back to the original name.

The method “rename” takes a single argument.



Detector alias (for referring to the detector within scripts)



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# BASIC ZAF CALCULATIONS

Create a material object

```
1> ss309=material("SS309")
2> zaf(ss309,d2,15.0)
```

Calculate the ZAF correction for the material “ss309”  
for the detector “d2” at “15.0” keV.

Material SS309 = [Cr(25.00 wgt%), Fe(55.00 wgt%), Ni(20.00 wgt%), Σ=100.00, 5 g/cc]

Detector Bruker 5 eV/ch - FWHM[Mn K $\alpha$ ]=127.9 eV - Sep 1, 2009 12:00:01 AM

Algorithm XPP - Pouchou & Pichoir Simplified

MAC NIST-Chantler 2005

E0 15 keV

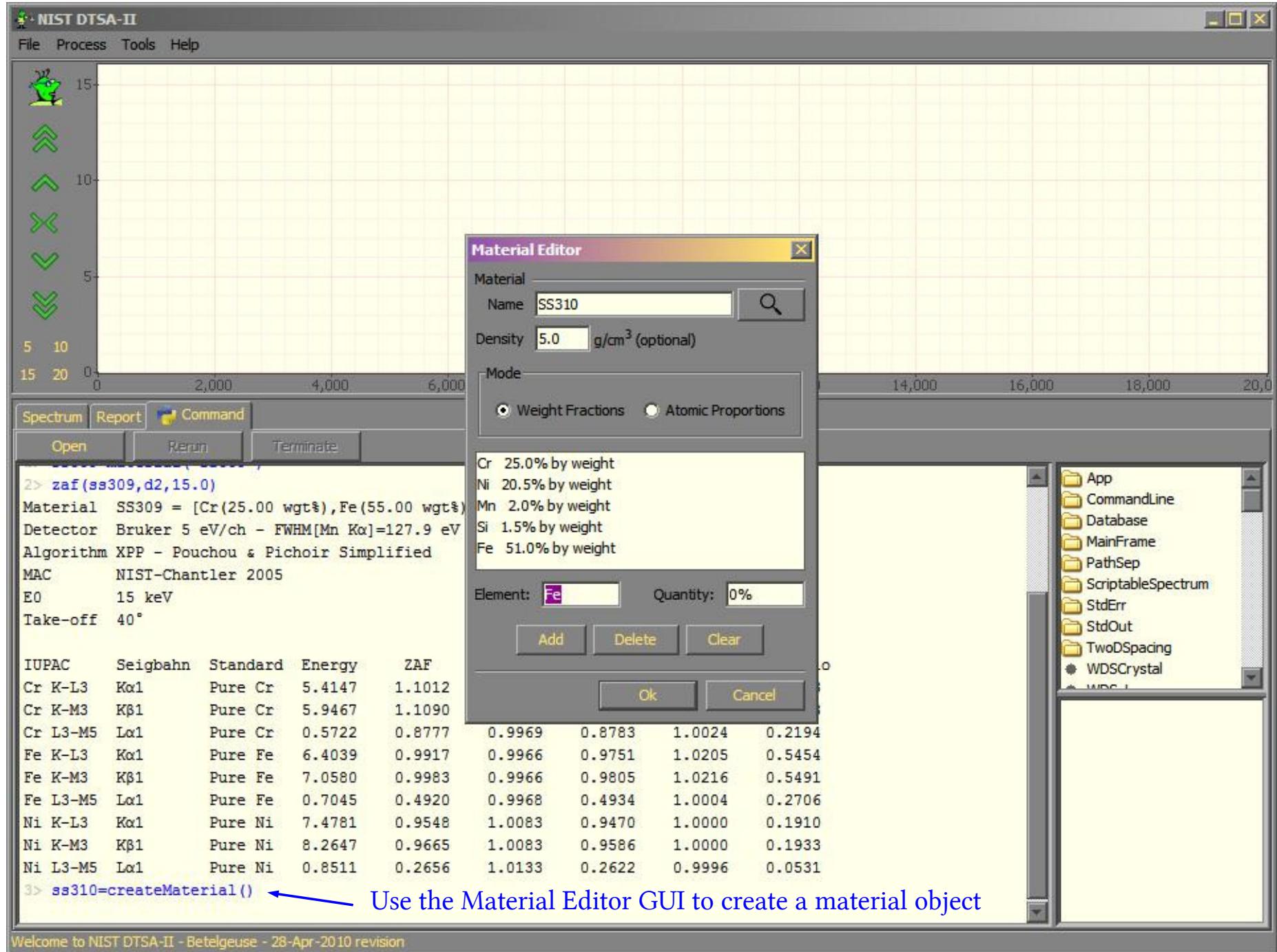
Take-off 40°

IUPAC	Seigbahn	Standard	Energy	ZAF	Z	A	F	k-ratio
Cr K-L3	K $\alpha$ 1	Pure Cr	5.4147	1.1012	0.9997	0.9939	1.1084	0.2753
Cr K-M3	K $\beta$ 1	Pure Cr	5.9467	1.1090	0.9997	0.9952	1.1147	0.2773
Cr L3-M5	L $\alpha$ 1	Pure Cr	0.5722	0.8777	0.9969	0.8783	1.0024	0.2194
Fe K-L3	K $\alpha$ 1	Pure Fe	6.4039	0.9917	0.9966	0.9751	1.0205	0.5454
Fe K-M3	K $\beta$ 1	Pure Fe	7.0580	0.9983	0.9966	0.9805	1.0216	0.5491
Fe L3-M5	L $\alpha$ 1	Pure Fe	0.7045	0.4920	0.9968	0.4934	1.0004	0.2706
Ni K-L3	K $\alpha$ 1	Pure Ni	7.4781	0.9548	1.0083	0.9470	1.0000	0.1910
Ni K-M3	K $\beta$ 1	Pure Ni	8.2647	0.9665	1.0083	0.9586	1.0000	0.1933
Ni L3-M5	L $\alpha$ 1	Pure Ni	0.8511	0.2656	1.0133	0.2622	0.9996	0.0531

3>



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Use the Material Editor GUI to create a material object

Welcome to NIST DTSA-II - Betelgeuse - 28-Apr-2010 revision



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# ZAF AGAINST SPECIFIC STANDARDS

```
13> ss310=createMaterial()
14> zaf(ss310,d2,15.0,stds={"Fe":"SS309","Cr":"SS309","Ni":"SS309"})
Material SS310 = [Si(1.50 wgt%),Cr(25.00 wgt%),Mn(2.00 wgt%),Fe(51.00 wgt%),Ni(20.50 wgt%),Σ=100.00,5 g/cc]
Detector Bruker 5 eV/ch - FWHM[Mn Kα]=127.9 eV - Sep 1, 2009 12:00:01 AM
Algorithm XPP - Pouchou & Pichoir Simplified
MAC NIST-Chantler 2005
E0 15 keV
Take-off 40°
```

Defaults to pure elements.

Use the specified standard

IUPAC	Seigbahn	Standard	Energy	ZAF	Z	A	F	k-ratio
Si K-L3	K $\alpha$ 1	Pure Si	1.7397	0.7178	1.1516	0.6228	1.0008	0.0108
Si K-M3	K $\beta$ 1	Pure Si	1.8290	0.7532	1.1516	0.6535	1.0009	0.0113
Cr K-L3	K $\alpha$ 1	SS309	5.4147	0.9913	0.9977	0.9997	0.9939	0.9913
Cr K-M3	K $\beta$ 1	SS309	5.9467	0.9911	0.9977	0.9998	0.9936	0.9911
Cr L3-M5	L $\alpha$ 1	SS309	0.5722	0.9890	0.9980	0.9912	0.9999	0.9890
Mn K-L3	K $\alpha$ 1	Pure Mn	5.8987	1.0043	0.9776	0.9973	1.0302	0.0201
Mn K-M3	K $\beta$ 1	Pure Mn	6.4904	0.9762	0.9776	0.9734	1.0259	0.0195
Mn L3-M5	L $\alpha$ 1	Pure Mn	0.6332	0.4405	0.9764	0.4502	1.0021	0.0088
Fe K-L3	K $\alpha$ 1	SS309	6.4039	0.9983	0.9976	0.9999	1.0008	0.9257
Fe K-M3	K $\beta$ 1	SS309	7.0580	0.9965	0.9976	0.9982	1.0007	0.9240
Fe L3-M5	L $\alpha$ 1	SS309	0.7045	0.9527	0.9980	0.9546	1.0000	0.8834
Ni K-L3	K $\alpha$ 1	SS309	7.4781	0.9991	0.9975	1.0016	1.0000	1.0241
Ni K-M3	K $\beta$ 1	SS309	8.2647	0.9987	0.9975	1.0013	1.0000	1.0237
Ni L3-M5	L $\alpha$ 1	SS309	0.8511	1.0242	0.9980	1.0262	1.0000	1.0498

15>

Note: When quantified using a similar standard,  
the Z, A and F terms are all close to 1.

This is likely to produce accurate quantifications!!!



# TABULATING ELEMENTAL DATA

listEdges("Pb")

```
17> listEdges("Pb")
IUPAC      Siegbahn   Energy (keV)
Pb K        Pb K       88.0045
Pb L1       Pb LI      15.8608
Pb L2       Pb LII     15.2
Pb L3       Pb LIII    13.0352
Pb M1       Pb MI      3.8507
Pb M2       Pb MII     3.5542
Pb M3       Pb MIII    3.0664
Pb M4       Pb MIV     2.5856
Pb M5       Pb MV      2.484
Pb N1       Pb NI      0.8936
Pb N2       Pb NII     0.7639
Pb N3       Pb NIII    0.6445
Pb N4       Pb NIV     0.4352
Pb N5       Pb NV      0.4129
Pb N6       Pb NVI     0.1429
Pb N7       Pb NVII    0.1381
Pb O1       Pb OI      0.1473
Pb O2       Pb OII     0.1048
Pb O3       Pb OIII    0.086
Pb O4       Pb OIV     0.0218
Pb O5       Pb OV      0.0192
```

listTransitions("Fe")

```
18> listTransitions("Fe")
IUPAC      Siegbahn   Weight   Energy (keV)   Wavelength (Å)
Fe K-L3    Kα1        1         6.4039      1.93607
Fe K-L2    Kα2        0.5118    6.3909      1.94001
Fe K-M3    Kβ1        0.11895   7.058       1.75665
Fe K-M2    Kβ3        0.0684    7.058       1.75665
Fe K-M5    Kβ5        0.0001    7.1084      1.74419
Fe L3-M5   Lα1        1         0.7045      17.5989
Fe L3-M4   Lα2        0.1144    0.7045      17.5989
Fe L3-M1   Lℓ         0.10529   0.6152      20.1535
Fe L2-M4   Lβ1        0.16704   0.7175      17.28
Fe L2-M3   Lβ17       0.0001    0.6671      18.5855
Fe L2-N1   Ly5         0.0045    0.718       17.268
Fe L2-M1   Lη         0.0511    0.6282      19.7364
Fe L1-M3   Lβ3        0.0479    0.7921      15.6526
Fe L1-M2   Lβ4        0.02569   0.7921      15.6526
```

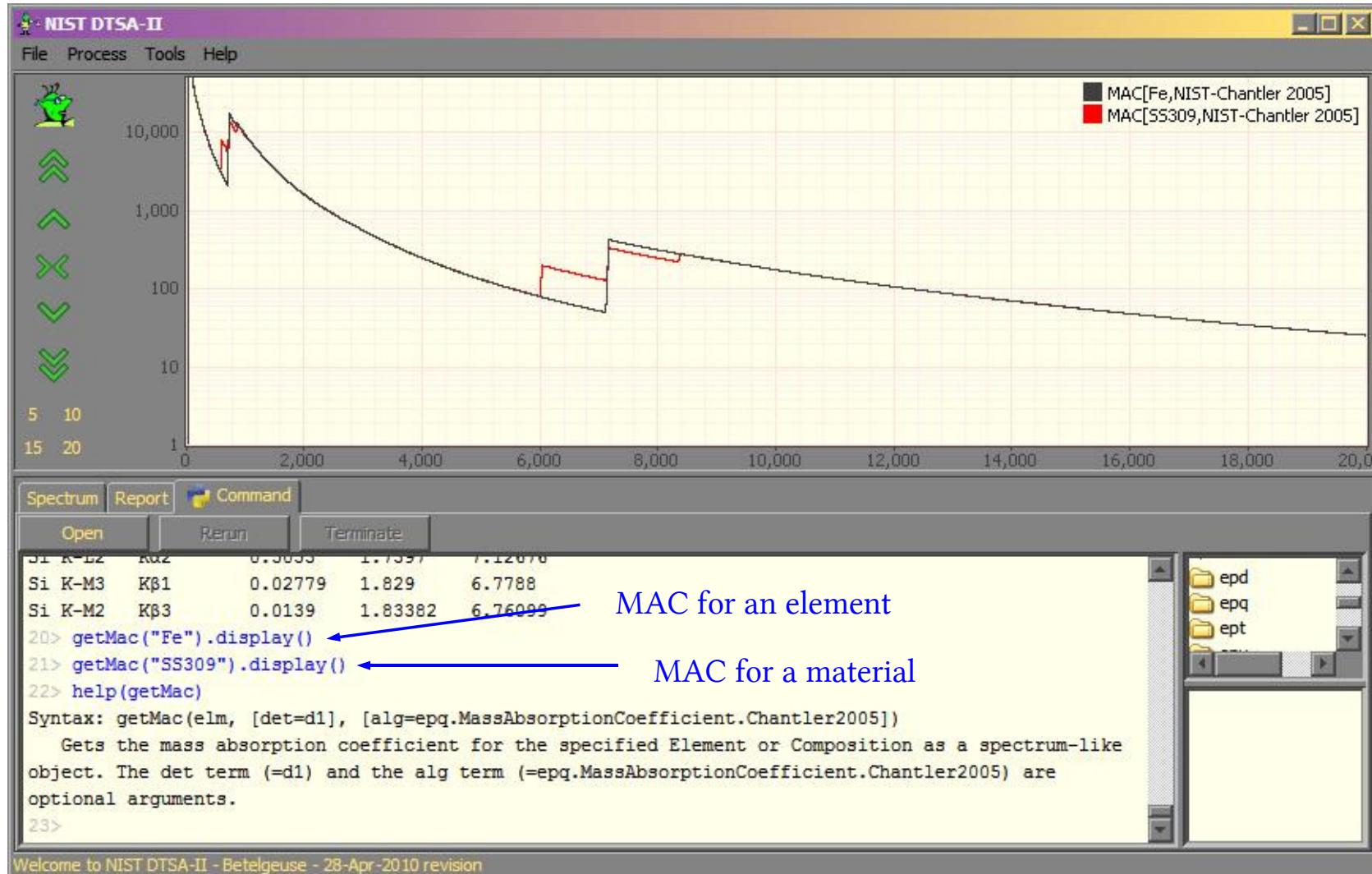
listData("Si")

```
19> listData("Si")
Abbrev   Name      Z      A
Si       Silicon   14     28.0855
IUPAC      Siegbahn   Energy (keV)
Si K       Si K      1.8389
Si L1      Si LI     0.1487
Si L2      Si LII    0.0992
Si L3      Si LIII   0.0992
Si M1      Si MI     0.0113572
Si M2      Si MII    0.00508305
IUPAC      Siegbahn   Weight   Energy (keV)   Wavelength (Å)
Si K-L3   Kα1        1         1.7397      7.12676
Si K-L2   Kα2        0.5053    1.7397      7.12676
Si K-M3   Kβ1        0.02779   1.829       6.7788
Si K-M2   Kβ3        0.0139    1.83382    6.76099
20>
```

listData(..) = listEdges(...) + listTransitions(...) + extra



# DISPLAYING MAC DATA



getMac(...) actually creates a spectrum-like object. Hence the call to the method “display()” plots the MAC on the spectrum window (which I’ve scaled to log10.)

# CALCULATING MAC DATA

```
23> help(mac)
Syntax: mac(mat, xx, [alg=epq.MassAbsorptionCoefficient.Chantler2005])
where mat is a material or element (by name, Material or Element object), xx is an element,
atomic shell or x-ray transition and alg is the tabulation to use (by default Chantler2005).
Displays the mass absorption coefficient for the specified element or x-ray transition in the
specified material.

24> mac("SS309","Fe")  
XRT      MAC  
          (cm^2/g)  
Fe K-L2   163.494  
Fe K-L3   162.578  
Fe K-M2   125.719  
Fe K-M3   125.719  
Fe K-M5   128.58  
Fe L1-M2  10924.9  
Fe L1-M3  10924.9  
Fe L2-M1  6774.49  
Fe L2-M3  5878.88  
Fe L2-M4  11044.9  
Fe L2-N1  11026.4  
Fe L3-M1  7117.92  
Fe L3-M4  5661.85  
Fe L3-M5  5661.85  
25> |
```

For all transitions associated with an element.

```
27> mac("SS309",transition("Fe L3-M5"))
XRT      MAC  
          (cm^2/g)  
Fe L3-M5  5661.85
```

For a specific transition in IUPAC notation.



# USEFUL OBJECTS

## Creating various types of useful objects:

- element: creates epq.Element objects
- atomicShell: creates epq.AtomicShell objects
- material: creates either epq.Composition or epq.Material objects.

```
29> elm=element("Fe")
30> shell=atomicShell("Fe L3")
31> xrt=transition("Fe L3-M5")
32> comp=material("Fe2O3")
33> mat=material("Fe2O3",5.24)
```

Many of the objects manipulated from the command line are actually Java objects.  
Many are from one of the libraries within *epq.jar*

The fully qualified name of the Element object is:

*gov.nist.microanalysis.EPQLibrary.Element*

This is a little verbose so within the scripting the alias:

*epq.Element*

is equivalent.

Common aliases:

<i>epq</i>	<i>gov.nist.microanalysis.EPQLibrary</i>
<i>epu</i>	<i>gov.nist.microanalysis.Utility</i>
<i>ept</i>	<i>gov.nist.microanalysis.EPQTools</i>
<i>epd</i>	<i>gov.nist.microanalysis.EPQLibrary.Detector</i>

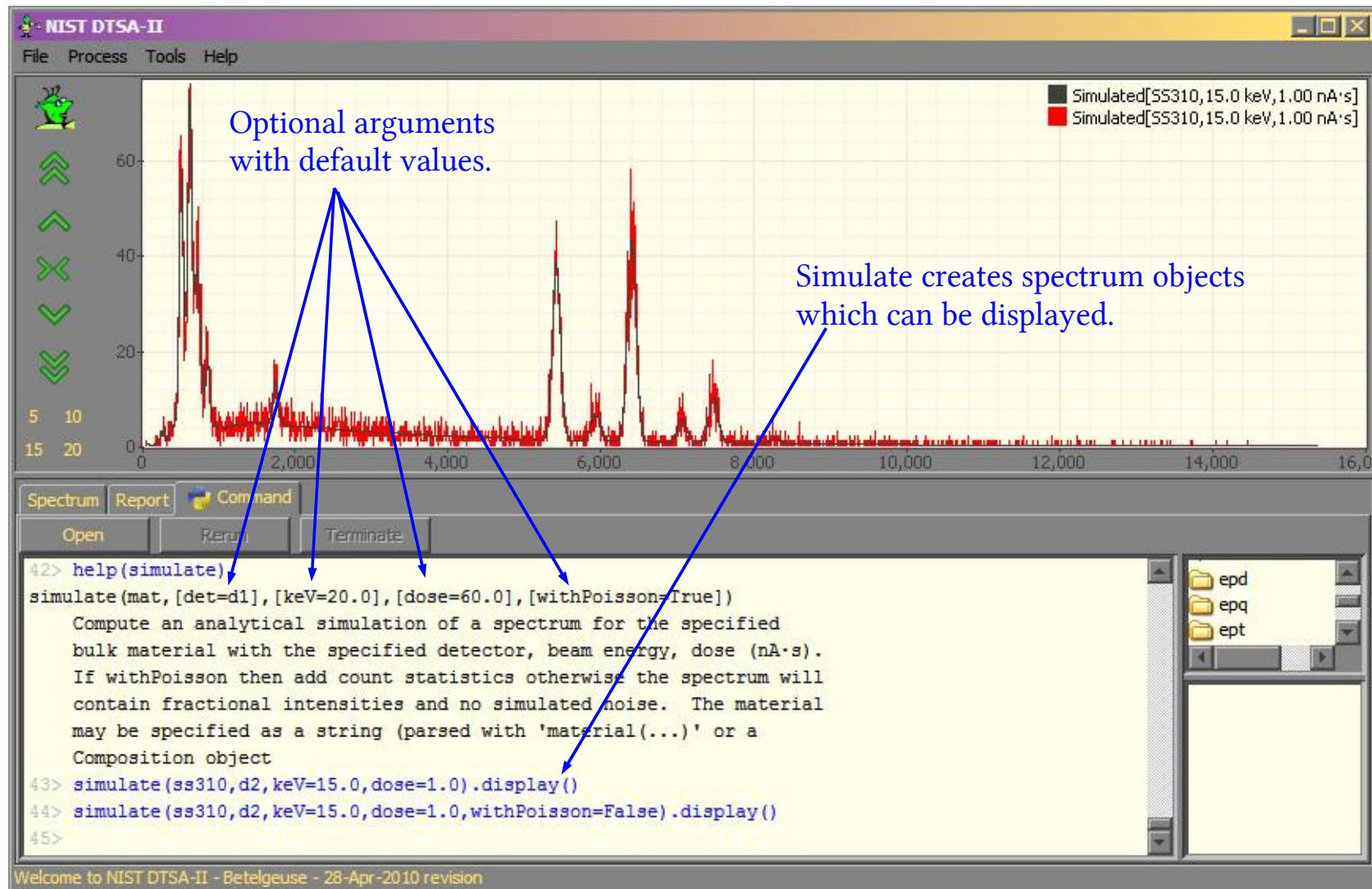
Library documentation:

<http://www.cstl.nist.gov/div837/837.02/epq/dtsa2/JavaDoc/index.html>

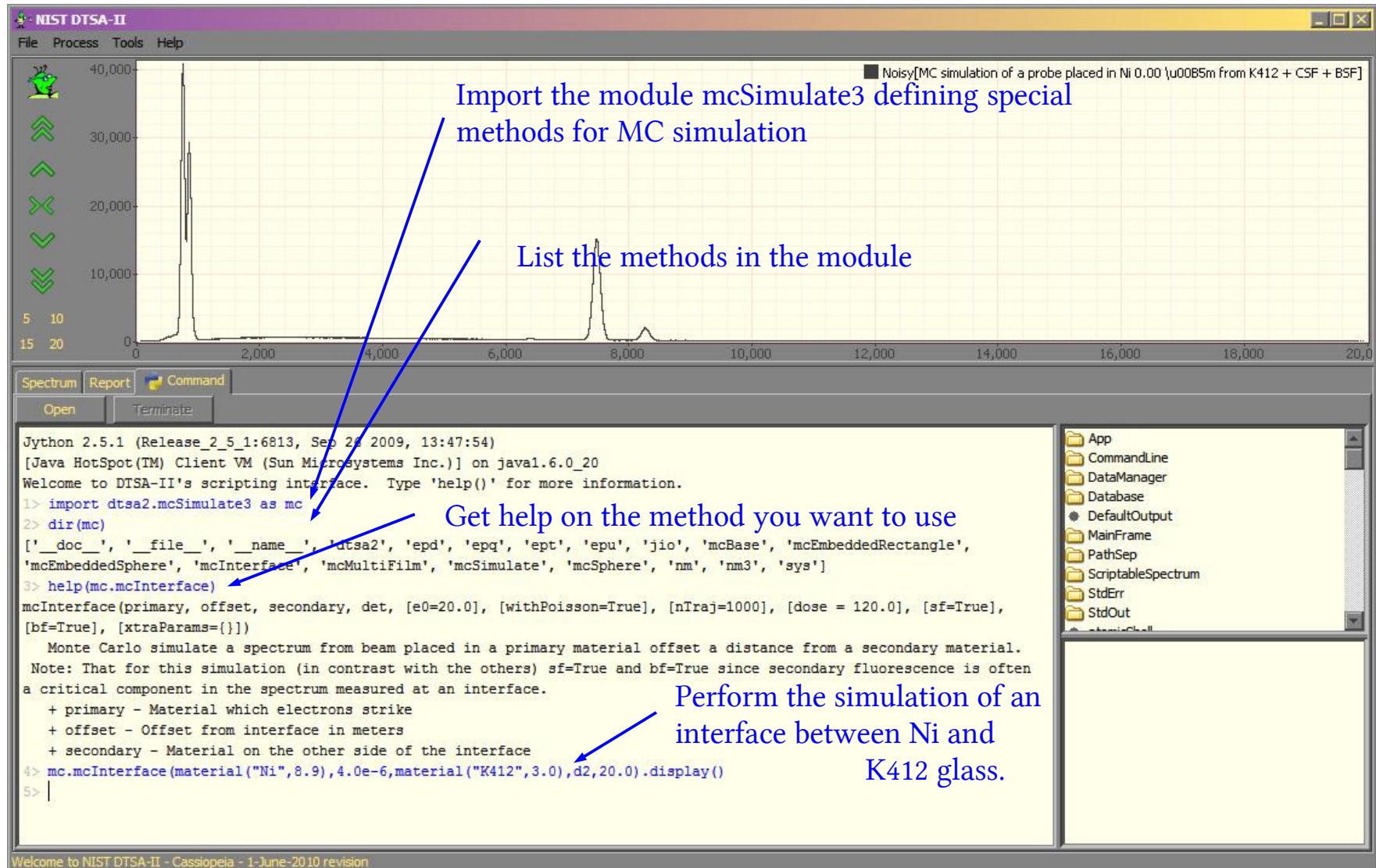


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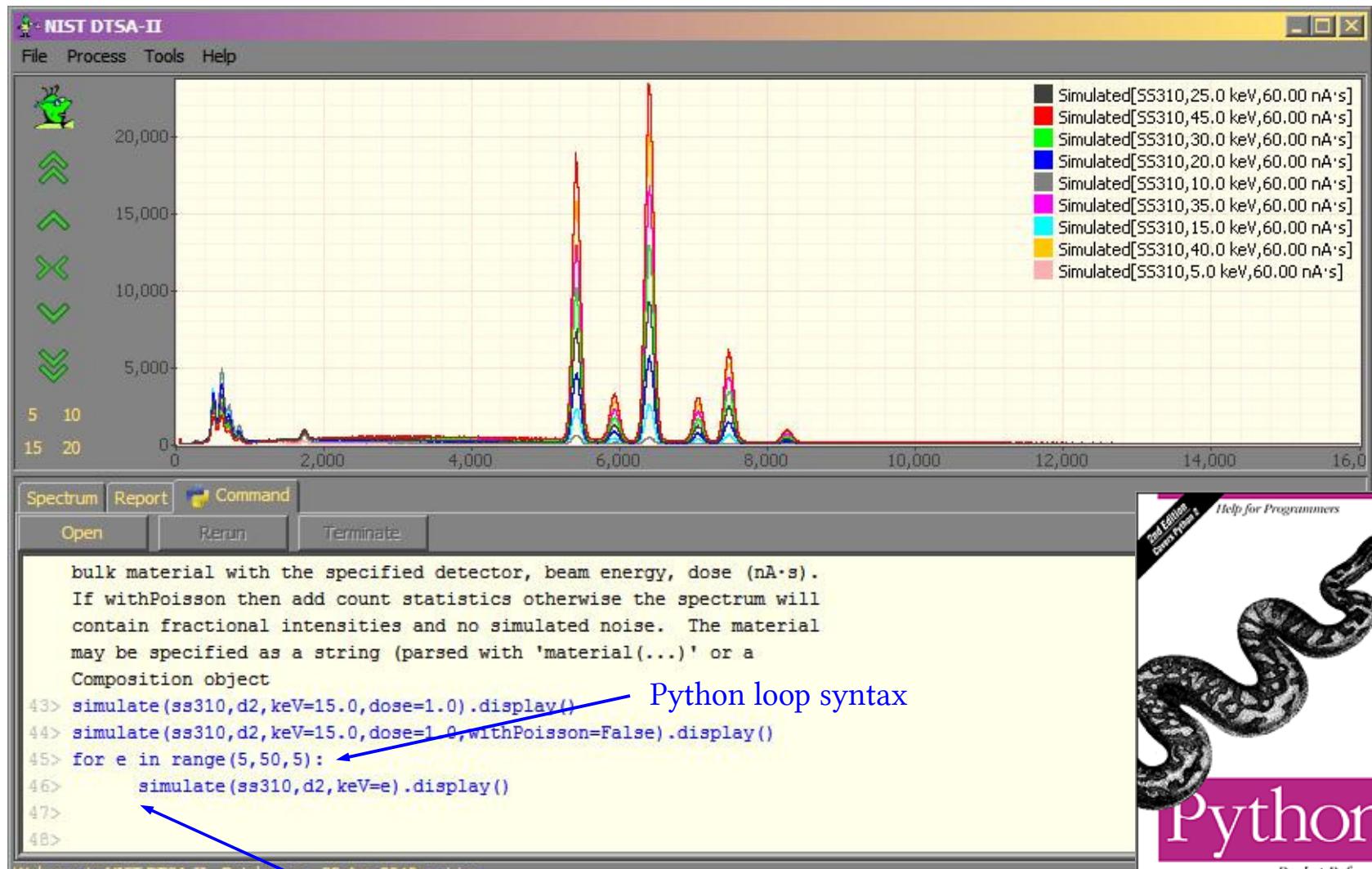
# QUICK SIMULATION



# MONTE CARLO SIMULATION



# LOOPS



Indentation dictates structure



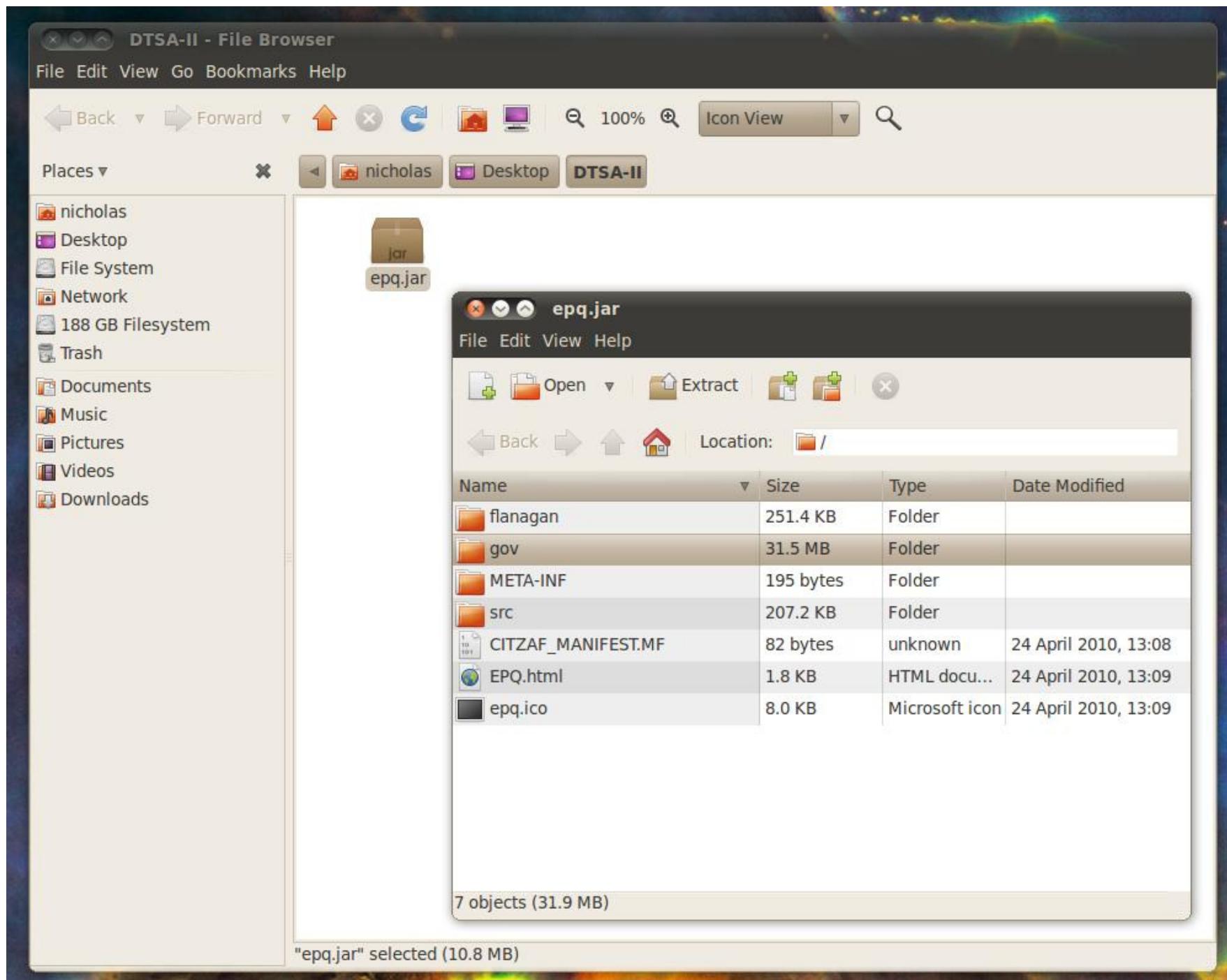
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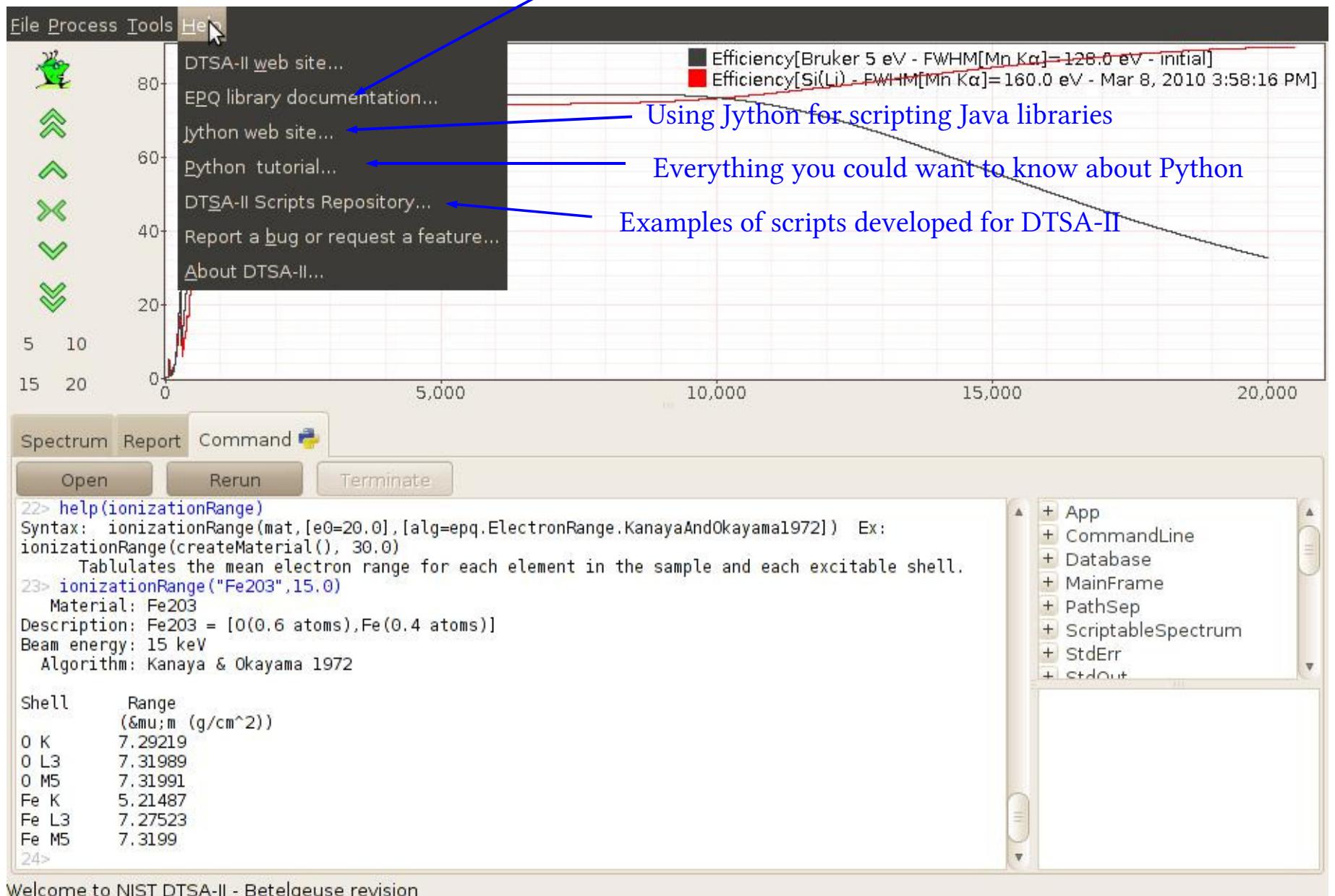
```
22> help(ionizationRange)
Syntax: ionizationRange(mat,[e0=20.0],[alg=epq.ElectronRange.KanayaAndOkayama1972]) Ex:
ionizationRange(createMaterial(), 30.0)
    Tablulates the mean electron range for each element in the sample and each excitable shell.
23> ionizationRange("Fe203",15.0)
    Material: Fe203
Description: Fe203 = [O(0.6 atoms),Fe(0.4 atoms)]
Beam energy: 15 keV
Algorithm: Kanaya & Okayama 1972

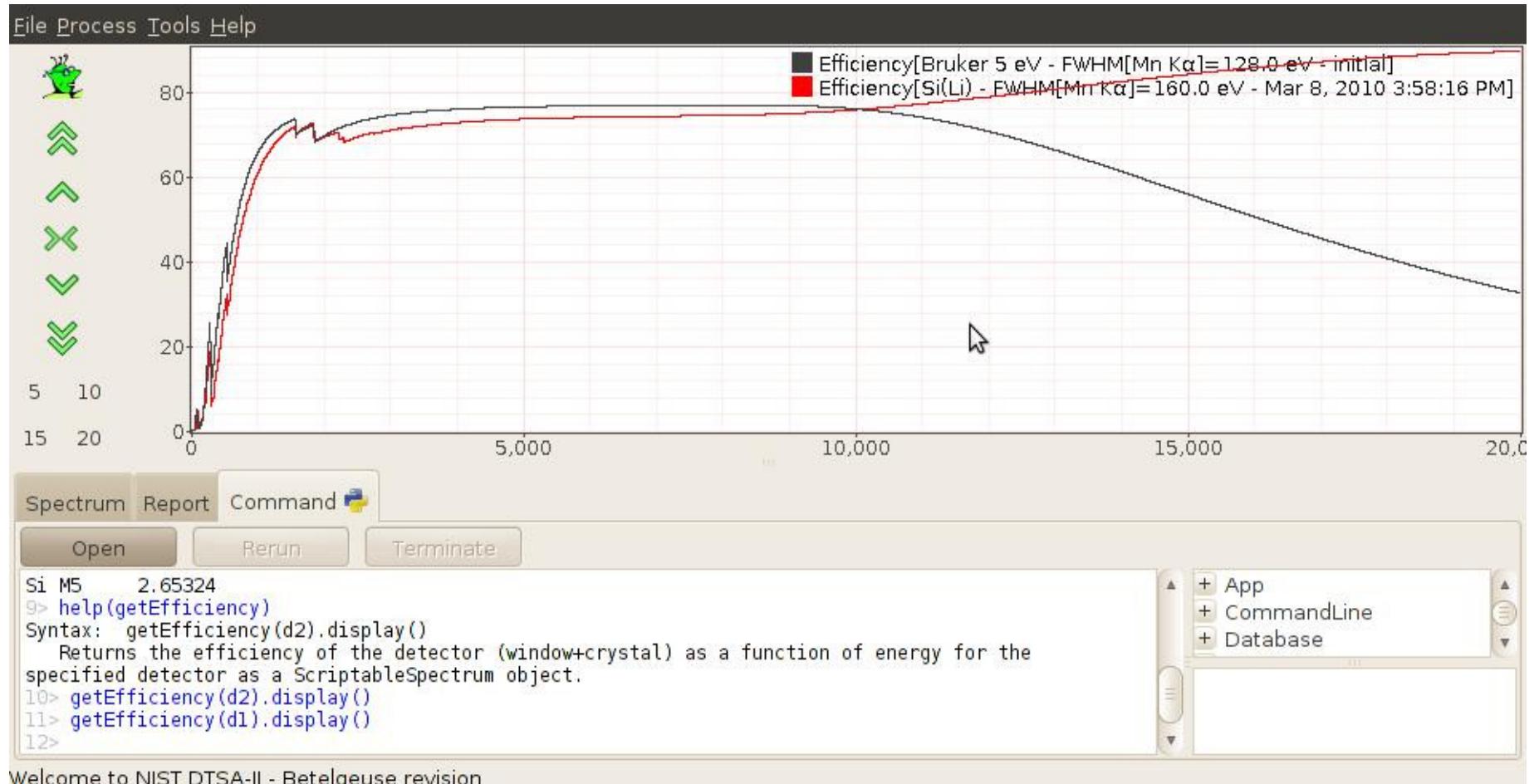
Shell      Range
(&mu;m (g/cm^2))
0 K        7.29219
0 L3       7.31989
0 M5       7.31991
Fe K       5.21487
Fe L3      7.27523
Fe M5      7.3199
24>
```

If you hand ionizationRange a Material (Composition + density) the output is in  $\mu\text{m}$  otherwise for Compositions it is in  $\mu\text{m} (\text{g}/\text{cm}^3)$ .









`getEfficiency(d2)` gets the effective quantum efficiency of the detector – includes window, support, dead layers, coatings, crystal geometry – as a spectrum-line object.

Notice how the SDD loses efficiency at high energies. This is due to the relative thinness of the detector (0.5 mm vs 3 mm for a conventional Si(Li)). Also observe how the



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tc.py \*

```
for z in range(6,85):
    mat = material(epq.Element.byAtomicNumber(z).toAbbrev())
    tmp = "%d" % z
    for e0 in range(10,50,10):
        tc=epq.SpectrumUtils.totalCounts(simulate(mat,d2,e0))
        tmp = "%s\t%g" % ( tmp, tc)
    print tmp
```

Python ▾

Tab Width: 3 ▾

Ln 6, Col 34

INS

